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the sum of the numbers of protons and neutrons in the nucleus is called the mass number. This is because each proton and each neutron weigh one atomic mass unit (amu). By adding together the number of protons and neutrons and multiplying by 1 amu, you can calculate the mass of the atom.

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The basic units in chemistry will be defined: atoms, molecules, subatomic particles. Then, we will discuss them from a beginner point of view, and formulate them in the format of questions. Therefore, we aim this article to people that are unfamiliar with chemistry or with science in general. If you are a teacher, you might want to redirect your students here. Disclaimer: This is not intended to be a comprehensive description of each concept, but rather an introduction to each of them ...

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Introduction to Basic Chemistry Chemistry is the science and study of matter, including its properties, composition as well as reactivity.

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Chemistry relates to everything that can be sensed from the minute elements to complex structures. The atom and molecules are the basic unit or components of Chemistry.

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NEET Chemistry Some Basic Concepts Of Chemistry Questions ...

'Some basic concepts of chemistry' is the most fundamental chapter of complete chemistry. It gives information about the atomic number and mass number of elements. In any chemical reaction, it

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is important for us to know about the number of reactants that will consume and the products that will produce, thus for estimating all these calculations, we use the laws of chemical combinations.

Some Basic Concepts in Chemistry- Laws, Theory, Concepts ...

1.2 Atomic Structure. Atoms are the basic building blocks of chemistry. This topic will provide students with a detailed understanding of how the sub-particles in an atom are arranged and will also look at the evidence found to give the more sophisticated picture of the atom model used in AS level chemistry.

Chemistry - CCEA

(a) A molecule of a compound has atoms of different elements. (b) A compound cannot be separated into its constituent elements by

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physical methods of separation. (c) A compound retains the physical properties of its constituent elements. (d) The ratio of atoms of different elements in a compound is fixed

Some basic concepts of chemistry Multiple Choice Question ...

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Some Basic Concepts of Chemistry – Test) , ()))) ? 32

7.0221367×10^{32} ; Question 57 of 106. 57. Question In Avogadro's law volume is directly proportional to _____. Pressure Number of moles ...

MCQs of Chemistry Class 11 Chapter 1 | Some Basic Con ...

This solution contains questions, answers, images, explanations of the complete chapter 1 titled Some Basic Concepts Of Chemistry of Chemistry taught in Class 11. If you are a student of Class 11 who is using NCERT Textbook to study Chemistry, then you must come across chapter 1 Some Basic Concepts Of Chemistry.

NCERT Solutions For Class 11 Chemistry Chapter 1 Some ...

The first question included in CBSE NCERT Solutions Chemistry

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Chapter 1 explains Some Basic Concepts of Chemistry Class 11 that requires you to calculate the molecular mass of compounds that are formed due to the chemical bonding of elements. As a beginner in the concepts of chemistry, you may find the calculation complex to understand.

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